

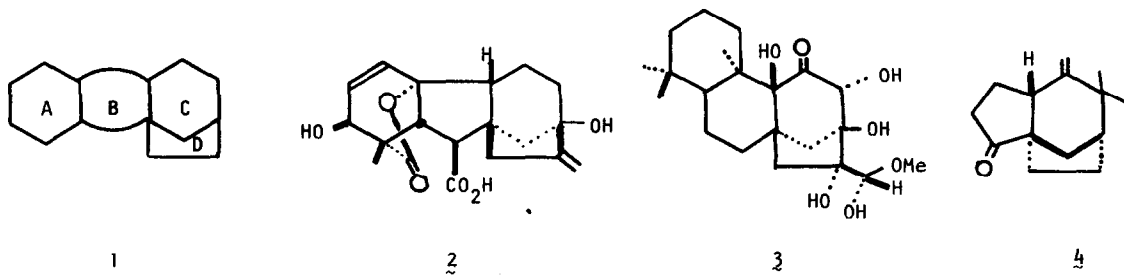
A NEW APPROACH TO THE BICYCLO[3.2.1]OCTANE RING SYSTEM:

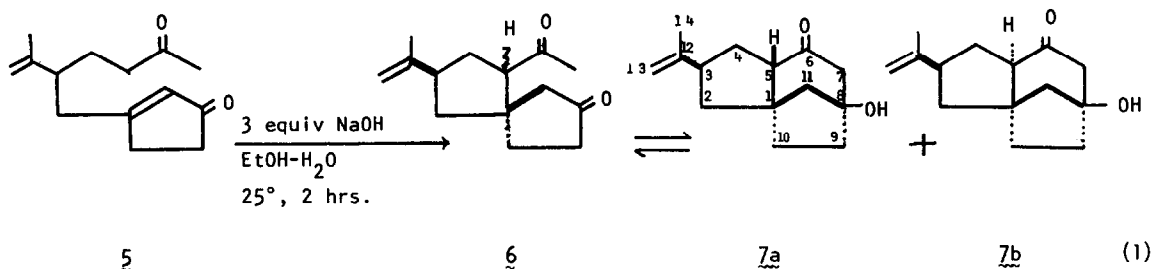
INTRAMOLECULAR MICHAEL AND ALDOL REACTIONS OF 3-(5-OXOHXYL)-2-CYCLOPENTENONES.

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Bicyclo[3.2.1]octanes fused to other rings (general structure 1) are a basic structural unit found in various important natural products. Examples include the plant-growth stimulator gibberelic acid (2),¹ the antibiotic phlebiakauranol (3),² and the fragrant norsesquiterpene khusimone (4).³ Synthetic strategies for construction of bicyclo[3.2.1]octanes fused to one other ring (i.e., rings BCD of structure 1) have usually involved sequential construction of the three rings, with the most common approach being attachment of ring D onto a preformed BC system.⁴ Coates' protocol involved attachment of ring B onto a preformed CD system,⁵ and Trost's recent approach utilized bicycloannulation of ring C onto a spiro BD ring system.^{4b} In each of these approaches a third ring is added to two existing carbocycles. A more convergent and biogenetically patterned strategy would involve elaboration of the BCD tricyclic array directly from a monocyclic precursor. Pattenden has suggested a photochemical variation of this convergent approach,⁶ and Buchi has successfully used a monocyclic 5-(4-hexenyl)-1,3-cyclohexadiene in an intramolecular Diels-Alder cyclization followed by an acid-catalyzed skeletal rearrangement to form bicyclooctane khusimone (4).³ Because of our interest in cyclopentanoid natural products,⁷ we had occasion to prepare 3-(5-oxohexyl)-2-cyclopentenone 5 via a simple and efficient route as described in the accompanying communication.⁸ We report here that monocyclic cyclopentenone 5 undergoes a mild, efficient, base-promoted, double cyclization to form a tricycle containing the bicyclo[3.2.1]octane ring system (eq. 1).





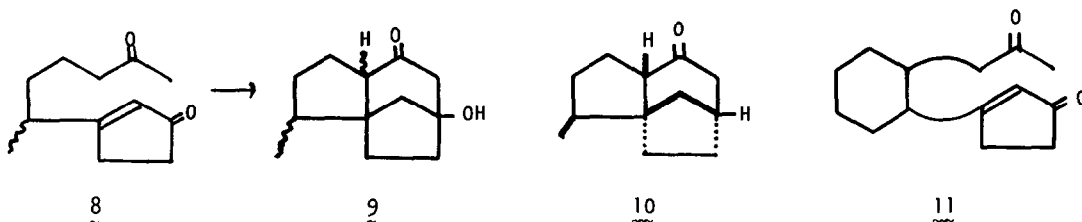
Exposing cyclopentenone 5 to sodium hydroxide in aqueous ethanol for 2 hours at 25° led to complete disappearance of the ir and nmr signals characteristic of cyclopentenone 5. Repeated fractional crystallization of the crude product from chloroform and petroleum ether gave crystalline bicyclo[3.2.1]octanone 7a and crystalline 7b in 38% and 11% yields, respectively. The mother liquor had spectral properties consistent with spiro-cyclopentanone 6 (e.g., a ^{13}C nmr signal at 213.23 ppm). Treating the mother liquor with sodium hydroxide in aqueous ethanol produced the same mixture of 6, 7a and 7b as was formed from monocyclic cyclopentenone 5. Likewise, ketols 7a and 7b when submitted to these conditions produced the same product distribution of 6, 7a and 7b. Thus the initial Michael addition appears to be irreversible for this system under these conditions, whereas the subsequent aldol condensation seems to be reversible.

The major product, mp 112-112.5°C, displayed the following spectral data which are consistent with structure 7a: ir (CHCl_3): 3600(m), 3430(b&m), 3080(w), 2995(sh), 2940(s), 2865(m), 1705(vs), 1640(m), 1450(m), 1375(m), 1350(m), 1320(m), 1280(m), 1155(m), 1135(m), 1070(m), 900cm^{-1} (s); ^{13}C nmr (CDCl_3): δ 207.25(s, C_6), 148.07(q, C_{12}), 108.54(t, C_{13}), 80.02(s, C_8), 60.56(d, C_3), 56.13(t, C_2), 50.89(s, C_1), 49.80(t, C_7), 43.13(d, C_5), 41.86(t, C_{11}), 37.34(t, C_9), 30.09(t, C_4), 25.64(t, C_{10}), 20.70(q, C_{14}); ^1H nmr (CDCl_3): δ 4.6(s, 2H), 2.85-2.4(m, 5H), 2.4-2.0(m, 2H), 1.95-1.3(m, 11H); Mass spectrum (70eV, rel intensity): 220(M^+ , 76), 43(100); Anal. Calcd for $\text{C}_{14}\text{H}_{20}\text{O}_2 \cdot \frac{1}{3}\text{H}_2\text{O}$: C, 73.32; H, 9.21. Found: C, 74.36; H, 9.02. This structural assignment was confirmed by x-ray measurements.⁹

The minor product (7b), mp 102-102.5°C, had very similar ir and ^1H nmr signals as those of the major product (7a). Very small differences in the ^{13}C nmr were evident: ^{13}C nmr (CDCl_3): δ 207.10(s, C_6), 148.01(q, C_{12}), 108.66(t, C_{13}), 80.08(s, C_8), 60.68(d, C_3), 56.27(t, C_2), 51.02(s, C_1), 49.96(t, C_7), 43.28(d, C_5), 42.04(t, C_{11}), 37.22(t, C_9), 30.22(t, C_4), 25.78(t, C_{10}), 20.70(q, C_{14}).

From the spectroscopic similarity of these two products and the known 2-4:1 equilibrium distribution of closely related hydrindanones,¹⁰ we concluded that bicyclo[3.2.1]octanes 7a and 7b were epimeric at C₅. Further support for the epimeric relationship at C₅ rather than at the isopropenyl-bearing C₃ center came from ¹³C nmr which showed that the isopropenyl groups in both 7a and 7b had very similar chemical shifts (the largest difference being 0.18 ppm).

Monocyclic 3-(5-oxohexyl)-2-cyclopentenone 8 likewise underwent a base-promoted intramolecular Michael addition followed by an internal aldol cyclization to form a product with spectral data consistent with tricyclic ketol 9. Ketol 9 can be converted¹¹ into bicyclo[3.2.1]octanone 10 which is the key intermediate in Coates' total synthesis of (+)-zizaene.^{5,12,13} The rapid, double cyclization of both methyl-substituted and isopropenyl-substituted 3-(5-oxohexyl)-2-cyclopentenones 8 and 5 suggests that this new approach may also be applicable to direct construction of tetracyclic arrays such as 1 from bicyclic precursors (e.g. 11).



We anticipate that application of this new, convergent, double cyclization approach will greatly facilitate total synthesis of various important bicyclo[3.2.1]octanes.

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References

1. For a review, see B. E. Cross *et. al.*, in "Gibberellins, Advances in Chemistry Series," vol. 28, American Chemical Society, Washington, 1961.
2. (a) T. C. McMorris and M. Anchel, Tetrahedron, 23, 3985 (1967); (b) J. M. Lisy, J. Clardy, M. Anchel, and S. M. Weinreb, J. Chem. Soc. Chem. Commun., 406 (1975).
3. G. Buchi, A. Hauser, and J. Limacher, J. Org. Chem., 42, 3323 (1977); *cf.* E. G. Breitholle and A. G. Fallis, *ibid.*, 43, 1966 (1978).
4. (a) For a review, see J. MacMillan and R.J. Pryce, Phytochem., 3, 283 (1973); (b) for a recent alternate approach with many leading references, see B. M. Trost and L. H. Latimer, J. Org. Chem., 43 1031 (1978).
5. R. M. Coates and R. L. Sowerby, J. Amer. Chem. Soc., 94, 5386 (1972).
6. M. Mellor, D. A. Otieno and G. Pattenden, J. Chem. Soc. Chem. Commun., 138 (1978).
7. (a) G. H. Posner, and C. M. Lentz, Tetrahedron Lett., 3215 (1977); and (b) G. H. Posner, J. J. Sterling, C. E. Whitten, C. M. Lentz, and D. J. Brunelle, J. Amer. Chem. Soc., 97, 107 (1975).
8. A. Alexakis, M. J. Chapdelaine, G. H. Posner and A. W. Runquist, Tetrahedron Lett., 000 (1978).
9. We thank Dr. Isabella Karle (U. S. Naval Research Laboratory, Washington) for the X-ray data, which will be published separately.
10. Hydrindanones: (a) M. Hanack, "Conformation Theory", Academic Press, New York, New York, 1965, pp 177-180; (b) C. S. Foote and R. B. Woodward, Tetrahedron, 20, 687 (1964). Hydroazulenones: (c) J. A. Marshall and W. F. Huffman, J. Amer. Chem. Soc., 92, 6358 (1970); (d) H. L. Goering, A. C. Olson and H. H. Epsy, *ibid.*, 78, 5371 (1956).
11. *cf.* (a) M. Mervic and E. Ghera, J. Amer. Chem. Soc., 99, 7673 (1977); (b) R. O. Hutchins, B. E. Maryonoff, and C. A. Milewski, J. Chem. Soc. Chem. Commun., 1097 (1971).
12. We thank Professor Coates for providing us with copies of his spectral data for various isomers of tricyclic ketone 10.
13. S. Inayama, T. Kawamata, and T. Ohkura, Tetrahedron Lett., 1557 (1978) report using an intramolecular aldol cyclization to link C₂ with C₇ in the pseudoguaiane skeleton of pulchellon.

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